

then submitted to paper chromatography using xylol as the developing solvent on Whatman No. 1 paper treated with formamide-acetone (1:3 by volume)⁸. The spots were marked through the ultraviolet absorption with the aid of fluorescence screen. Valine which has been shown to be N-terminal amino acid of trypsinogen both by dinitrofluorobenzene method² and Edman's

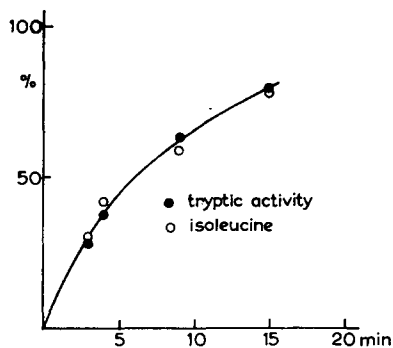


Fig. 1. Relationship between tryptic activity and N-terminal isoleucine.

to other amino acids¹¹.

Judging from these experimental results, it is likely that enterokinase acts as a peptidase-splitting valylpeptide³ from trypsinogen. The details will be published in *Acta Chem. Scand.* in 1956. This work was made possible through a research grant to Professor ERIK JORPES from the Swedish State Medical Research Council. The author's thanks are also due to Dr. PEHR EDMAN for his help during the performance of the work.

Chemistry Department II, Karolinska Institutet, Stockholm (Sweden)
and Department of Physiological Chemistry, University of Lund (Sweden)

IKUO YAMASHINA

¹ J. H. NORTHOP, M. KUNITZ AND R. M. HERRIOTT, *Crystalline Enzymes*, 2nd ed., Columbia Univ. Press, New York, 1948, p. 125.

² M. ROVERY, C. FABRE AND P. DESNUELLE, *Biochim. Biophys. Acta*, 9 (1952) 702; *ibid.*, 10 (1953) 481; *ibid.*, 12 (1953) 547.

³ E. W. DAVIE AND H. NEURATH, *J. Biol. Chem.*, 212 (1955) 515.

⁴ J. H. NORTHOP, M. KUNITZ AND R. M. HERRIOTT, *Crystalline Enzymes*, 2nd ed., Columbia Univ. Press, New York, 1948, p. 262.

⁵ I. YAMASHINA, *Arkiv Kemi*, 7 (1954) 539.

⁶ I. YAMASHINA, *ibid.*, in the press.

⁷ P. EDMAN, *Acta Chem. Scand.*, 4 (1950) 277; *ibid.*, 4 (1950) 283.

⁸ P. EDMAN, unpublished.

⁹ I. YAMASHINA, unpublished.

¹⁰ P. EDMAN AND O. E. ARVIDSSON, unpublished.

¹¹ A. LEVY AND C. H. LI, *J. Biol. Chem.*, 217 (1955) 355.

Received January 17th, 1956

Mitochondrial oxidative phosphorylation in a magnesium-free medium

The assumption, that magnesium is a cofactor in the reactions involved in oxidative phosphorylation^{1,2,3}, is based on the fact that maximum P/O values for mitochondrial preparations could be obtained only in media containing added Mg^{++} . An assembly of respiratory enzymes, capable of oxidative phosphorylation without added Mg^{++} , has recently been obtained by LEHNINGER's group⁴. Its endogenous magnesium content is unknown.

With intact rat liver mitochondria, using the rapid platinum electrode technique for simultaneous measurement of respiratory and phosphorylative effects⁵, we have obtained respiratory control and maximum P/O values for the oxidation of β -hydroxybutyrate or succinate in a medium containing neither magnesium ions nor other divalent cations. The mitochondria were prepared as in ref.⁶. Table I shows P/O values for β -hydroxybutyrate oxidation in a magnesium-containing (I) and a magnesium-free (II) medium.

TABLE I

P/O VALUES FOR THE OXIDATION OF
 β -HYDROXYBUTYRATE

Experiment	Medium	P/O
1	I*	2.8
	II**	2.9
2	I	2.8
	II	2.8

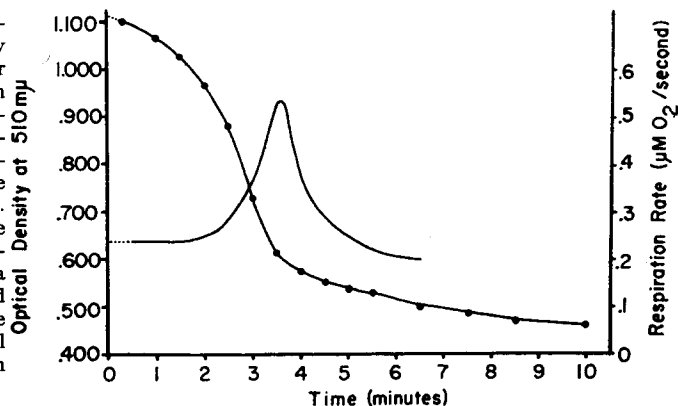
* 16 mM phosphate buffer, 12 mM sodium fluoride, 6 mM magnesium chloride and 84 mM potassium chloride.

** 16 mM phosphate buffer, 12 mM sodium fluoride and 93 mM potassium chloride. The pH of I and II was 7.2.

An uncoupling event of a kind not previously reported occurs some minutes after incubation in medium II at room temperature with β -hydroxybutyrate as substrate. Uncoupling is here defined as loss of respiratory control. The event is manifested by a temporary, two- to fourfold increase of the respiration rate. During this interval the ability of adenosine diphosphate and 2,4-dinitrophenol to stimulate the respiration disappears. The uncoupling event can be correlated with a rapid fall in optical density at 510 m μ , due to swelling and structural disorganization of the mitochondria⁶. The controls in medium I had a constant respiration rate and optical density at 510 m μ . As the fall occurs somewhat before the uncoupling event (see Fig. 1) one can postulate that the morphological change causes the respiratory stimulation and not *vice versa*. With succinate the changes appear later and adenosine phosphates have a stabilizing effect: the changes occur after a period several times longer and are more gradual. Thus, especially for succinate oxidation, in a magnesium-free medium containing an adenosine phosphate the stability should be great enough to permit maximum P/O values even with the Warburg technique. This was indeed recently observed by ERNSTER AND LÖW⁷.

Our results show that the lack of Mg⁺⁺ in the medium causes a great and significant morphological change in the particles and indicate that a role of magnesium in oxidative phosphorylation of isolated mitochondria is to maintain the proper mitochondrial structure. The question whether magnesium also takes part in the actual oxidative phosphorylation reactions remains open.

Fig. 1. Change of respiration rate and optical density at 510 m μ with time for mitochondria suspended in medium II. Zero time: addition of 10 mM (final concentration) of β -hydroxybutyrate 30 seconds after the mitochondria were added. Temperature: 25° C. In the measurement of the respiration rate the mitochondria were twice as concentrated as in the measurement of the optical density. ● = optical density, — = respiration rate.



Many thanks are due to Dr. BRITTON CHANCE for his support and stimulating advice.

The Johnson Foundation for Medical Physics,
The University of Pennsylvania School of Medicine,
Philadelphia, Pa. (U.S.A.)

HERRICK BALTSCHIEFFSKY*

¹ A. L. LEHNINGER, *The Harvey Lectures*, 1953-1954, Academic Press Inc., New York, 1955, p. 176.

² S. H. MUDD, J. HARTING PARK AND F. LIPMANN, *Proc. Natl. Acad. Sci. U.S.*, 41 (1955) 571.

³ H. A. LARDY, *3rd Internat. Congr. Biochem., Brussels, 1955, Conférences et Rapports*, p. 287.

⁴ C. COOPER, T. M. DEVLIN AND A. L. LEHNINGER, *Biochim. Biophys. Acta*, 18 (1955) 159.

⁵ B. CHANCE AND G. R. WILLIAMS, *Nature*, 175 (1955) 1120.

⁶ J. RAAFLAUB, *Helv. Physiol. Pharmacol. Acta*, 11 (1953) 142.

⁷ L. ERNSTER AND H. LÖW, *Exptl. Cell Research, Suppl.* 3, (1955) 133.

Received February 20th, 1956

* Supported by grants from "Ella och Georg Ehrnrooth's stipendiefond" and the Lallor Foundation.